

3,5-Dibromo-2,2,6,6-tetramethylpiperidin-4-one

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Key indicators

Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$
 R factor = 0.043
 wR factor = 0.090
Data-to-parameter ratio = 17.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the title compound, $\text{C}_9\text{H}_{15}\text{Br}_2\text{NO}$, the piperidin-4-one ring
adopts a chair conformation.Received 23 January 2007
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Comment

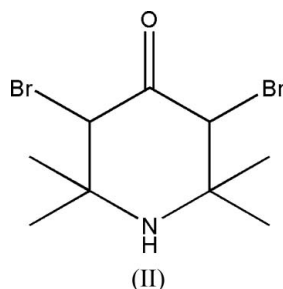
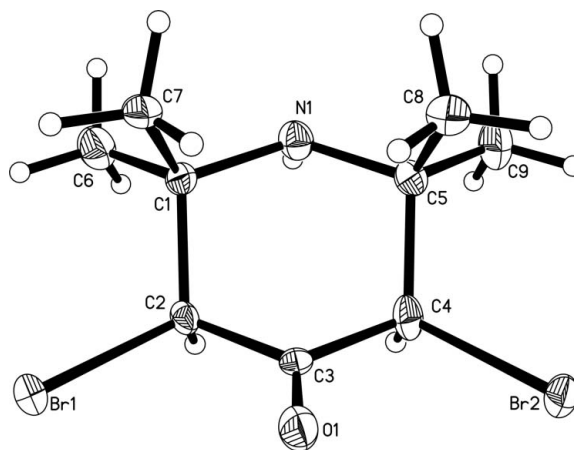
2,2,6,6-Tetramethylpiperidin-4-one, (I), plays an important
role as a starting material in the research and development of
HALS (hindered amine light stabilizers) (Yoshihiko *et al.*,
2004). In our studies on HALS, many derivatives of (I) have
been prepared. The title compound, (II) (Fig. 1), is one of
these compounds and its structure is reported here.In the molecule of (II), the bond lengths and angles are
within normal ranges (Allen *et al.*, 1987). The piperidin-4-one
ring adopts a chair conformation. The N1–H1a group partici-
pates in a very weak intermolecular hydrogen bond with
atom O1 of an adjacent molecule (Table 1); this is possible
because of its axial location with respect to the six-membered
ring.

Figure 1
The molecular structure of (II), showing 50% displacement ellipsoids
(arbitrary spheres for the H atoms).

Experimental

The title compound was synthesized by the reaction of 2,2,6,6-tetramethylpiperidin-4-one with bromine in the presence of acetic acid (Couet *et al.*, 1985). Colourless prisms of (II) were obtained by recrystallization from an acetone solution (m.p. 418 K). Analysis calculated for C₉H₁₅Br₂NO: C 34.53, H 4.83, Br 51.05, N 4.47%; found: C 34.79, H 4.56, N 4.38%.

Crystal data

C ₉ H ₁₅ Br ₂ NO	Z = 4
M _r = 313.04	D _x = 1.866 Mg m ⁻³
Monoclinic, P2 ₁ /c	Mo Kα radiation
a = 8.8677 (18) Å	μ = 7.24 mm ⁻¹
b = 21.968 (4) Å	T = 293 (2) K
c = 6.0078 (12) Å	Prism, colourless
β = 107.78 (3)°	0.34 × 0.18 × 0.16 mm
V = 1114.5 (4) Å ³	

Data collection

Rigaku Saturn diffractometer	8984 measured reflections
ω scans	2181 independent reflections
Absorption correction: multi-scan (Jacobson, 1998)	1667 reflections with I > 2σ(I)
T _{min} = 0.210, T _{max} = 0.314	R _{int} = 0.079
	θ _{max} = 26.0°

Refinement

Refinement on F ²	H atoms treated by a mixture of independent and constrained refinement
R[F ² > 2σ(F ²)] = 0.043	w = 1/[σ ² (F _o ²) + (0.0326P) ²]
wR(F ²) = 0.090	where P = (F _o ² + 2F _c ²)/3
S = 0.99	(Δ/σ) _{max} = 0.001
2181 reflections	Δρ _{max} = 0.84 e Å ⁻³
125 parameters	Δρ _{min} = -0.86 e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1a···O1 ⁱ	0.86 (4)	2.64 (4)	3.488 (6)	169 (6)

Symmetry code: (i) x, y, z + 1.

The N-bound H atom was located in a difference map and its position was refined freely with U_{iso}(H) = 1.5U_{eq}(N). Other H atoms were positioned geometrically (C—H = 0.96–0.98 Å) and refined as riding, with U_{iso}(H) = 1.2U_{eq}(C) or 1.5U_{eq}(methyl C).

Data collection: *CrystalClear* (Rigaku/MSK, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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